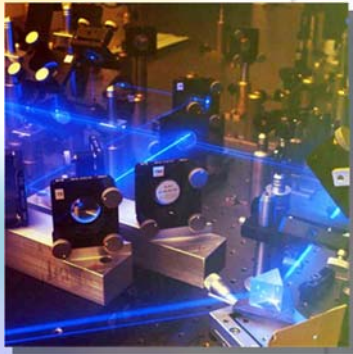


Physics Academy

Laser Physics

Introduction to Laser Essentials

Lecture 1



Dr. Hazem Falah Sakeek

www.hazemsakeek.com

www.physicsacademy.org

Things you need to know

Before studying about lasers, you must be familiar with **basic terms** used to describe electromagnetic waves:

Wavelength (λ)

Frequency (ν)

Period (T)

Velocity of light (c)

Index of refraction (n)

We will **briefly review** these terms, but it is much better if you are familiar with:

Some terms from **geometric optics** such as: **refraction**, **reflection**, **thin lenses** etc.

Some terms from "**Modern Physics**" such as **photons**, **Models of atoms**, etc.

Electromagnetic Radiation

Electromagnetic Radiation is a **transverse wave**, advancing in vacuum at a constant speed which is called: **velocity of light**.

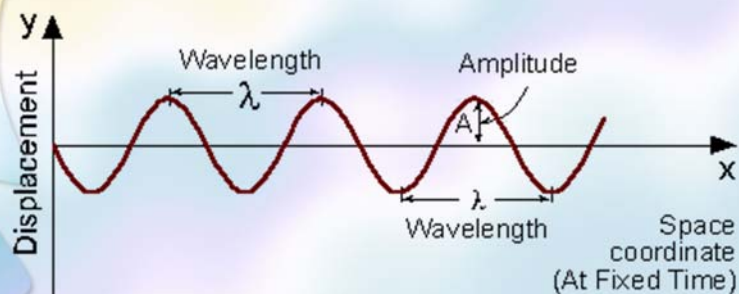
All electromagnetic waves have the same velocity in vacuum, and its value is approximately:

$$c = 300,000 \text{ [km/sec]} = 3 \times 10^8 \text{ [m/sec]}$$

One of the most important parameters of a wave is its **wavelength**.

Wavelength

Wavelength (λ) (Lamda) is the distance between two adjacent points on the wave, which have the same **phase**. As an example (see figure below) the distance between two adjacent peaks of the wave.



Frequency

In a parallel way it is possible to define a wave by its **frequency**.

Frequency (μ) is defined by the **number of times that the wave oscillates per second**.

Between these two parameters the relation is:

$$c = \lambda * \mu$$

From the physics point of view, **all electromagnetic waves are equal (have the same properties) except for their wavelength (or frequency)**.

As an example: the speed of light is the same for visible light, radio waves, or x-rays.

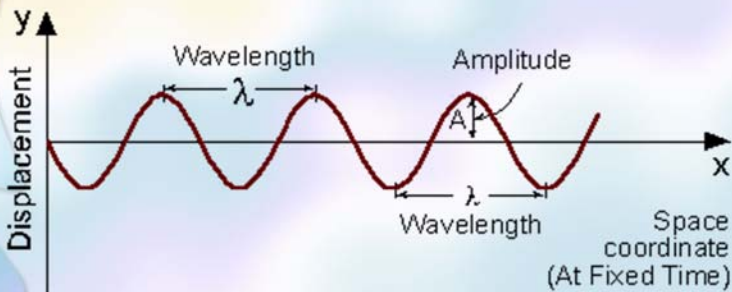
Wave Description

A **wave** can be described in two standard forms:

1. **Displacement as a function of space when time is held constant.**
2. **Displacement as a function of time at a specific place in space.**

Displacement as a function of space

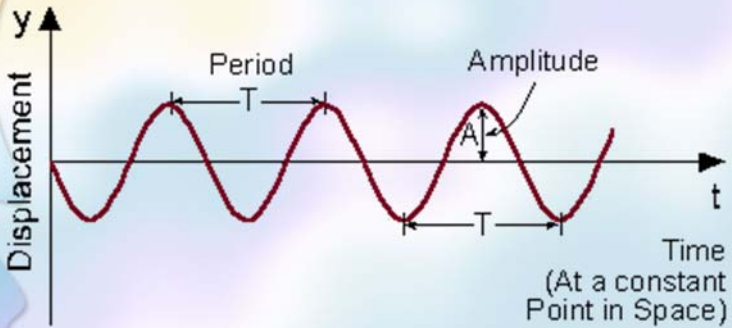
Displacement as a function of space, when time is "frozen" (held constant). In this description, **the minimum distance between two adjacent points with the same phase is wavelength (λ)**. Note that the horizontal (x) axis is **space coordinate**



A = Amplitude = Maximum displacement from equilibrium.

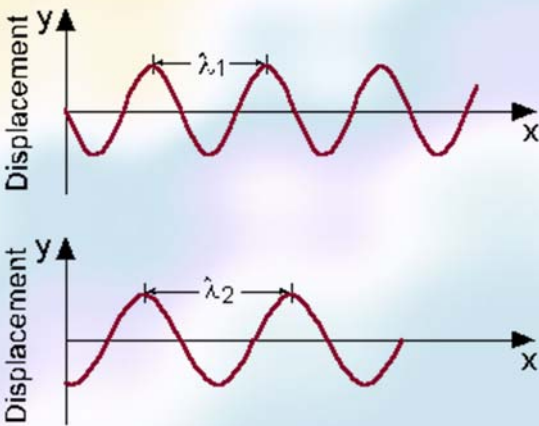
Displacement as a function of time

Displacement as a function of time, in a specific place in space, as described in figure. In this description, **the minimum distance between two adjacent points with the same phase is period (T)**. Note that the horizontal (x) axis is **time coordinate**



Wavelengths Comparison

The Figure describes how two different waves (with different wavelengths) look at a specific moment in time. Each of these waves can be uniquely described by its wavelength.



Photon Energy	Wavelength	Frequency	Common Name For the Spectral Region
E	λ	ν	
[eV]	[μm]	[Hz]	
$E = h\nu = \frac{hc}{\lambda} = \frac{h}{T}$	$\lambda = \frac{c}{\nu} = cT$	$\nu = \frac{c}{\lambda} = \frac{E}{h} = \frac{1}{T}$	
10 ³	10 ⁻³	10 ¹⁷	γ Rays
100	0.01	10 ¹⁶	X-Rays
10	0.1	10 ¹⁵	UV= Ultra-Violet
1	0.4 0.7	10 ¹⁴	Visible Spectrum
0.1	1	10 ¹³	IR= Infra-Red
0.01	10	10 ¹²	
10 ⁻³	10 ³	10 ¹¹	Microwave
10 ⁻⁴	10 ⁴	10 ¹⁰	Radio Waves

Violet
Blue
Green
Yellow
Orange
Red

The electromagnetic spectrum

The most important ideas summarized in figure are:

1. Electromagnetic waves **span** over many orders of magnitude in wavelength (or frequency).
2. The **frequency** of the electromagnetic radiation is inversely proportional to the wavelength.
3. The **visible spectrum** is a very small part of the electromagnetic spectrum.
4. **Photon energy** increases as the wavelength decreases. The shorter the wavelength, the more energetic are its photons.

Examples for electromagnetic waves are:

- **Radio-waves** which have wavelength of the order of **meters**, so they need big antennas.
- **Microwaves** which have wavelength of the order of **centimeters**. As an example: in a **microwave oven**, these wavelengths can not be transmitted through the protecting metal grid in the door, while the **visible spectrum** which have much shorter wavelength allow us to see what is cooking inside the microwave oven through the protecting grid.
- **x-Rays** which are used in medicine for taking pictures of the bone structure inside the body.
- **Gamma Rays** which are so energetic, that they cause ionization, and are classified as ionizing radiation.

Electromagnetic Radiation in Matter

Light Velocity in Matter

When electromagnetic radiation passes through matter with **index of refraction n** , its **velocity (v)** is less than the **velocity of light in vacuum (c)**, and given by the equation:

$$v = c / n$$

This equation is used as a **definition of the index of refraction**

$$n = (\text{speed of light in vacuum}) / (\text{speed of light in matter})$$

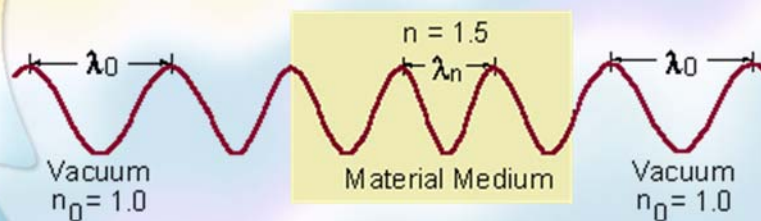
$$n = c / v$$

Gases, including air, are usually considered as having index of refraction equal to vacuum **$n_0=1$** .

The values of the index of refraction of most materials **transparent in the visible spectrum** is between **1.4-1.8**, while those of materials transparent in the Infra-Red (IR) spectrum are higher, and are **2.0-4.0**.

Wavelength in Matter

We saw that the velocity of light in matter is slower than in vacuum. This slower velocity is associated with reduced wavelength: $\lambda = \lambda_0 / n$, while the **frequency remains the same**



Refraction of Light Beam - Snell Law

Reducing the velocity of light in matter, and reducing its wavelength, causes **refraction of the beam of light**.

While crossing the border between two different materials, the light changes its direction of propagation according to the **Snell Equation**

$$n_1 \cdot \sin(\Theta_1) = n_2 \cdot \sin(\Theta_2)$$

Example

The velocity of Red light ($\lambda_0 = 0.6 \mu\text{m}$) in a certain medium is $1.5 \cdot 10^8 \text{ m/s}$. What is the wavelength of this light in this material?

Solution:

First find the index of refraction:

$$n = \frac{c}{v} = \frac{3 \cdot 10^8 \cdot \frac{\text{m}}{\text{s}}}{1.5 \cdot 10^8 \cdot \frac{\text{m}}{\text{s}}} = 2.0$$

Using n , calculate the wavelength in the material:

$$\lambda_n = \frac{\lambda_0}{n} = \frac{0.6 \cdot \mu\text{m}}{2.0} = 0.3 \cdot \mu\text{m}$$

Conclusion: The wavelength of Red light in a material with an index of refraction of 2.0, is $0.3 \mu\text{m}$